M. L. Wolfrom, Walter D. Harris, George F. Johnson, J. E. Mahan, Sam M. Moffett and Bernard Wildi. Osage Orange Pigments. XI. Complete Structures of Osajin and Pomiferin.
Page 406. In formula I, $b^{\prime}$ should be $b$, and $b$ should be $b^{\prime}$.

Page 410. In column 2, line 18, insert "3.4-dihydro-" before "11-( $p-$. ."-M. L. Wolfrom.
A. Bailey and H. M. Brooks. Electrolytic Oxidation of Lignin.

Page 446. In column 2, line 38, for "tn. p. 130.0$130.5^{\circ}$ " read " $93.8^{\circ}$."-A. Bailiey.
E. S. Lutton. The Identity and Polynorphism of Oleyldistearin from Kokum Butter.
Page 679. The last line of column 2 should read "Ivorydale 17, Ohio" instead of "Chapel Hill, N. C."

Murray Senkus. Reaction of Some Cyclic Acetals with Acid Anhydrides.

Page 735. In Table I, columin 2, line 4 should read "hexatrediol." Lines 14 and 16 should read "pentanediol." and the same correction in the last text line of column 2. and in line 7 . of the Summary on page 736 .

Page 736. In column 1, line 14 , for " 1,5 " read ${ }^{\prime} 1,6$ ' and in the last text line of col. 1 , for " 1,5 ' read ' 2,5 ." Merray Senkus.

Robert O. Sauer, W. J. Scheiber and Stuart D. Brewer. Derivatives of the Methylchlorosilanes. V. Polysiloxanes from Methyldichlorosilane.

Page 963. Addendum to Table III.
Value of $n$
1
2
2
3
4

| 760 mm. | B. p. ${ }^{\circ} \mathrm{C}$. |
| :---: | ---: |
| 142 | 10 mm. |
| 177 | 61.6 |
| 206 | 85.6 |
| $\ldots$ | 107 |

-Robert O. Sauer.
John E. Kilpatrick and Kenneth S. Pitzer. The Thermodynamics of 2,2 -Dimethylbutane.
Page 1071. In Table NII, line 2 should have " 3.547 ", instead of " 3.658 " and line 3 ' 7.276 " instead of " 7.160 ." This leaves the total unchanged. The height of the potential barriers is now 4260 instead of 4375 cal./mole (line 11 above Table XII).--K. S. Pitzer.

Walter M. Lauer, Richard T. Arnold, Burris Tiffany and John Tinker. The Synthesis of Some Chloromethoxyquitrolines.

Page 1268. In column 2, after line 27 , insert:
Anal. Calcd. for $\mathrm{C}_{\mathrm{tu}} \mathrm{H}_{8} \mathrm{ONCl}: \mathrm{C}, 62.04$ : H, t. 16 . Found: C, 62.37 ; H., 3.95.

A sample heated with hydrobromic acid gave the corresponding 4-chloro-7-laydroxyquinoline (m. p. 216-218 ${ }^{\circ}$.... W. M. Laver.

[^0]Ernest L. Jackson. 4-Amino-4'-hydroxylaminodiphenyl Sulfone, its Acetyl and d-Glucosyl Derivatives.
Page 1438. In column 1, line 14 , for "aminidophenyl" read "alninodiphenyl."

Robert C. Elderfield, et al. Alkylaminoalkyl Derivatives of 8-Aminoquinoline.

Page 1526. In the $R_{8}$ column (3), there should be a subscript 2 at the end of the $\mathrm{R}_{8}$ group formula for compounds 5, 19 and 44.-Robert C. Elderfield.

Donald R. Douslin and Hugh M. Huffman. Low-Temperature Thermal Data on the Five Isomeric Hexanes.

Page 1705. In Table II, line below the title, the molecular weight of hexane should be " 86.172 " instead of "84.176."
Page 1708. In Table VII the centered line " $n$-Hexane" above " 2 -Methylpentane" over the second section of the table, should have been placed above the first section, replacing the small type heading.-H. M. HuFFmas.

George Scatchard, Scott E. Wood and John M. Mochel. Vapor-Liquid Equilibrium. VII. Carbon TetrachlorideMethanol Mixtures.

Page 1960. In column two below Table I, what should have been the fifth line was accidentally omitted during correction and should read "benzene-methanol systemi it was not possible to."

Robert H. Baker and Arthur H. Schlesinger. The Reduction of Amide Vinylogs.
Page 2009. In column 1, the last two lines, for " 5 acetyldihydrouracil" read " 2 -Keto-1,2,3,4-tetrahydro- 5 -carbethoxy-6-methylpyrimidine.'-Robert H. Baker.

Louis A. Pinck and Guido E. Hilbert. Behavior of Certain Fluorene Compounds Containing Trivalent Carbon and Tetravalent Nitrogen.

Page 2011. Formula III should be


Louis A. Pinck.
Louis A. Pinck and Guido E. Hilbert. The Michael Condensation of Fluorene with Unsaturated Compounds.

Page 2015. The fourth formula should be Br


Br
Page 2016. In column 1, lines 10 and 11 should read: ${ }^{\circ}$ Calcd. for $\mathrm{C}_{26} \mathrm{H}_{12} \mathrm{Br}_{4}: \mathrm{C}, 48.46 ; \mathrm{H}, 1.86$. Found: C , $50.59 ;$ H, 2.12."-Louis A. Pinck.

DeLos F. DeTar and Marvin Carmack. The Willgerodt Reaction. II. A Study of Reaction Conditions with Acetophenone and Other Ketones.
Page 2025. The product in the first equation should be $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{CH}_{2} \mathrm{CONH}_{2} \cdots$ Marvin Carmack.


[^0]:    Albert S. Keston, Sidney Udenfriend and R. Keith Cannan. Mieroanalysis of Mixtures (Amino Acids) in the Forni of Isotopic Derivatives.

    Page 1390. In column 1, text line 12, read "trace"' instead of "trade." In columin 2, line 21, read " $7.5 \%$." In line 24, read "one-tenth" instead of "one-hundredth." Atbert S. Keston

